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(54) **ORGANIC/POLYMER
ELECTROLUMINESCENT DEVICES
EMPLOYING SINGLE-ION CONDUCTORS**

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(30) **Foreign Application Priority Data**

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(51) **Int. Cl.**
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(52) **U.S. Cl.** **428/690**; 428/917; 313/504; 313/506

(58) **Field of Classification Search** 428/690, 428/917; 257/40; 313/503, 504, 506
See application file for complete search history.

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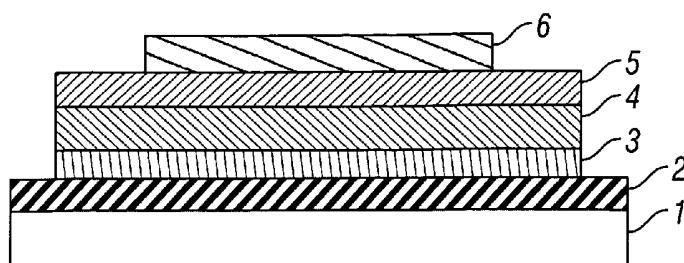
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(57) **ABSTRACT**

The present invention relates to electroluminescent devices employing single-ion conductors as the materials for an electron- or hole-injecting layer. Preferred electroluminescent devices employ an electron- or hole-injecting layer made of single-ion conductors in a conventional electroluminescent device which comprises: a transparent substrate; a semitransparent electrode deposited on the transparent substrate; a hole-injecting layer positioned on the semitransparent electrode; an electroluminescent layer made of organic luminescent material, positioned on the hole-injecting layer; an electron-injecting layer positioned on the electroluminescent layer; and, a metal electrode deposited on the electron-injecting layer. The electroluminescent devices of the invention have excellent electroluminescent efficiency and low turn-on voltage, which make possible their application to the development of high efficiency electroluminescent devices.

14 Claims, 1 Drawing Sheet



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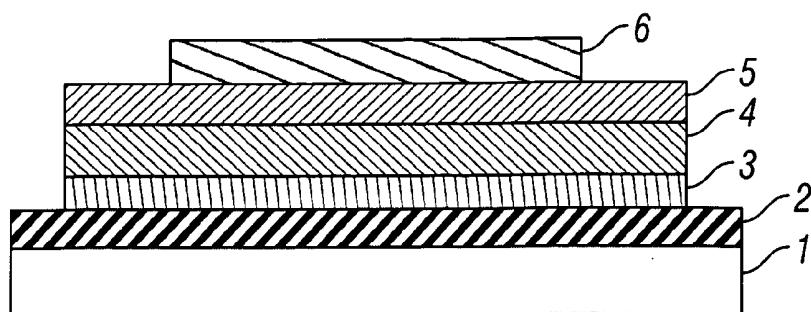


FIG. 1

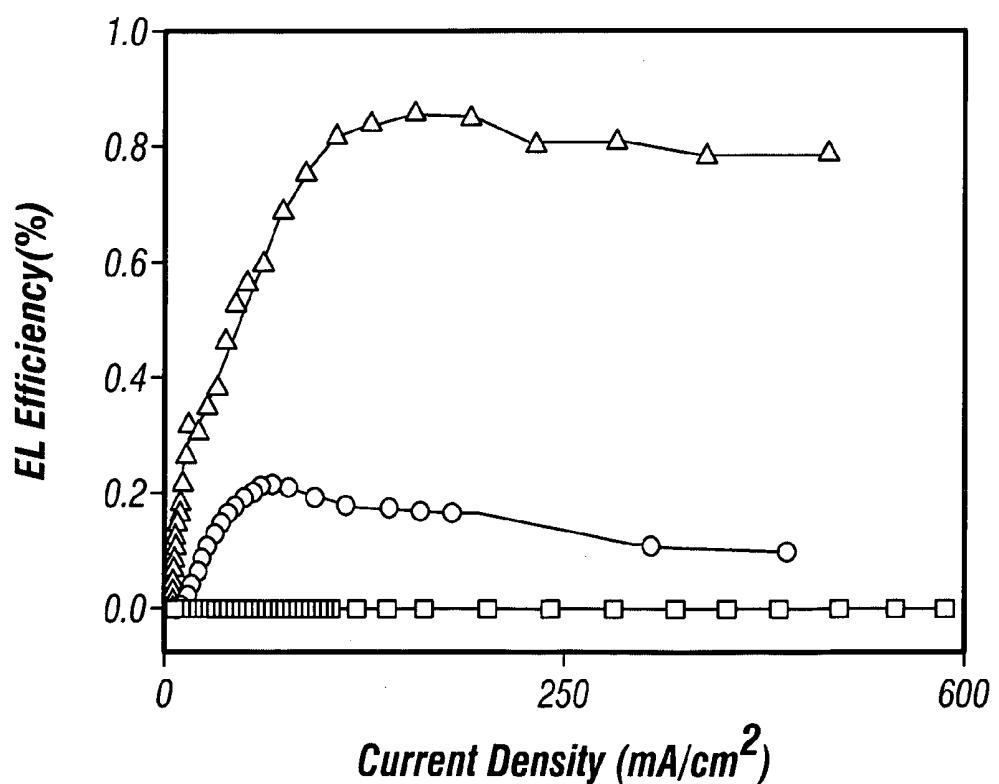


FIG. 2

**ORGANIC/POLYMER
ELECTROLUMINESCENT DEVICES
EMPLOYING SINGLE-ION CONDUCTORS**

RELATED APPLICATION INFORMATION

This application is a continuation under 35 U.S.C. § 365(c) claiming the benefit of the filing date of PCT application Ser. No. PCT/KR01/00535 designating the United States, filed 30 Mar. 2001 and published in English as WO 01/78464 A1 on 18 Oct. 2001, and which claims the benefit of the earlier filing date of Korean Patent Application No. 2000/16456, filed 30 Mar. 2000. The publication WO 01/78464 A1 is incorporated herein by reference.

BACKGROUND OF THE INVENTION

1. Field of the Invention

The present invention relates to electroluminescent devices employing single-ion conductors, more specifically, to organic/polymer electroluminescent devices employing single-ion conductors as an electron- or hole-injecting layer.

2. Description of the Related Art

Electroluminescent ("EL") devices that emit light by applying an electric field to the device typically comprise an ITO (indium tin oxide) substrate, EL material and two electrodes. To improve the EL efficiency, the device is provided with a hole-injecting layer between the ITO electrode and EL material, an electron-injecting layer between EL material and the counter metal electrode, or both layers. As the EL material that plays a crucial role in the device, organic polymer/inorganic hybrid nanocomposite employing insulating inorganic materials, such as SiO_2 and TiO_2 that help the transport of electric charges, has been developed and put to the practical use (see: S. A. Carter, *Applied Physics Letters*, 71:1145, 1997; L. Gozano, *Applied Physics Letters*, 73:3911, 1998).

In the meantime, studies on the hole- or electron-injecting layer have been actively performed to improve the EL efficiency, mainly by way of inserting ionomers as the electron-injecting layer (see: Hyang-Mok Lee et al., *Applied Physics Letters*, 72, 2382, 1998). However, it cannot be a basic solution to improve the EL efficiency because the movement of ions is restricted in the ionomers, which naturally limits electron-injection. As an alternative means for efficient electron-injection, an electron-transporting layer rather than the electron-injecting layer, was proposed in the art, which utilizes the materials that well transport electrons and have high affinity to the electrons. Several methods that utilize inorganic nanoparticles, 2-(4-biphenyl)-5-(4-tert-butylphenyl)-1,3,4-oxadiazole (PBD), or metal chelate complexes have been presented until now (see: U.S. Pat. Nos. 5,537,000; 5,817,431; and 5,994,835). However, these methods have not been realized in practical use due to the low EL efficiency or the difficulties confronted in the thin film deposition process.

Under the circumstances, there are strong reasons for developing and exploring a material that can be used as the hole- or electron-injecting layer to improve the EL efficiency while employing the convenient thin-film deposition process such as a spin-coating method.

SUMMARY OF THE INVENTION

The present inventors made an effort to develop a material that can improve the EL efficiency with convenient thin-film deposition process, and discovered that EL devices employ-

ing single-ion conductors as an electron- or hole-injecting layer show a highly improved EL efficiency.

A primary object of the present invention is, therefore, to provide EL devices employing single-ion conductors as an electron- or hole-injecting layer.

BRIEF DESCRIPTION OF THE DRAWINGS

The above, the other objects and features of the invention will become apparent from the following descriptions given in conjunction with the accompanying drawings, in which:

FIG. 1 is a schematic diagram showing a cross-sectional view of an organic/polymer EL device employing single-ion conductors of the present invention.

FIG. 2 is a graph showing the EL efficiency of an organic/polymer EL device employing a single-ion conductor as the electron-injecting layer, an organic/polymer EL device employing an ionomer as the electron-injecting layer, and an organic/polymer EL device without the electron-injecting layer.

Explanation of major parts of the drawings:

- 1: transparent substrate
- 2: semitransparent electrode
- 3: hole-injecting layer
- 4: electroluminescent layer
- 5: electron-injecting layer
- 6: metal electrode

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

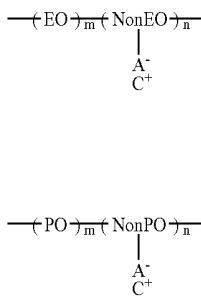
The organic/polymer EL device of the invention is improved in a sense that it employs electron- or hole-injecting layer made of single-ion conductors in a conventional EL device which comprises: a transparent substrate; a semitransparent electrode deposited on the transparent substrate; a hole-injecting layer positioned on the semitransparent electrode; an emissive layer made of organic luminescent material, positioned on the hole-injecting layer; an electron-injecting layer positioned on the emissive layer; and, a metal electrode deposited on the electron-injecting layer. The transparent substrate includes glass, quartz or PET (polyethylene terephthalate), and the semitransparent electrodes includes ITO (indium tin oxide), PEDOT (polyethylene dioxythiophene) or polyaniline.

The organic EL material includes: emissive conjugated polymers such as poly(para-phenylvinylene), poly(thiophene), poly(para-phenylene), poly(fluorene) or their derivatives; emissive non-conjugated polymers with side chains substituted with emissive functional groups such as anthracene; metal chelate complex of ligand structure such as emissive alumina quinone (Alq_3); low molecular-weight emissive organic material (monomers or oligomers) such as rubrene, anthracene, perylene, coumarine 6, Nile red, aromatic diamine, TPD (N,N'-diphenyl-N,N'-bis-(3-methylphenyl)-1,1'-biphenyl-4,4'-diamine), TAZ (3-(4-biphenyl)-4-phenyl-5-(4-tert-butylphenyl)-1,2,4-triazole) or other emissive monomeric or oligomeric material of the derivative of those material; laser dyes such as DCM (dicyanomethylene)-2-methyl-6-(p-dimethylaminostyryl)-4H-pyran), and blends of poly(meta-methylacrylic acid), polystyrene and poly(9-vinylcarbazole) with above-mentioned emissive materials. And, aluminum, magnesium, lithium, calcium, copper, silver, gold, or an alloy thereof is preferably employed for the metal electrode.

As the single-ion conductors, the materials containing ether chains ((— CH_2)_nO—) such as polyethylene oxide or

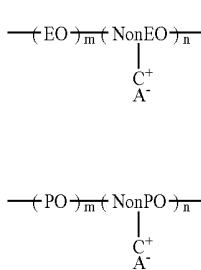
polypropylene oxide, and ionic groups such as SO_3^- , COO^- , I^- , or $(\text{NH}_3)_4^+$ in the main chains that form ionic bonds with counter ions such as Na^+ , Li^+ , Zn^{2+} , Mg^{2+} , Eu^{3+} , COO^- , SO_3^- , I^- , or $(\text{NH}_3)_4^+$ are preferably employed.

In general, single-ion conductors are classified into single-cation conductors (see: general formula (I), general formula (II)) and single-anion conductors (see: general formula (III) and general formula (IV)).



wherein EO represents ethyleneoxide; NonEO represent non-ethyleneoxide; PO represents propyleneoxide; NonPO represents non-propyleneoxide; A^- represents an anion; C^+ represents a cation; $m+n=1$; and, n represents a real number more than 0 and less than 1.

As shown in the general formula (I) and the general formula (II), single-cation conductors contain ether chains ($(-\text{CH}_2)_n\text{O}-$) such as polyethyleneoxide or polypropyleneoxide in the main chains, and anionic groups such as SO_3^- , COO^- , or I^- in the main or side chains which form ionic bonds with metal ions such as Na^+ , Li^+ , Zn^{2+} , Mg^{2+} , or Eu^{3+} , or other organic ions such as $(\text{NH}_3)_4^+$ as the counter ion.



wherein EO represents ethyleneoxide; NonEO represents non-ethyleneoxide; PO represents propyleneoxide; NonPO represents non-propyleneoxide; A^- represents anion; C^+ represents cation; $m+n=1$; and, n represents a real number more than 0 and less than 1.

As shown in general formula (III) and general formula (IV), single-anion conductor contains ether chains ($(-\text{CH}_2)_n\text{O}-$) such as polyethyleneoxide or polypropyleneoxide in the main chains, and cationic group such as $(\text{NH}_3)_4^+$ or $(-\text{CH}_2)_n\text{O}^+$ in the main or side chains which form ionic bonds with anions such as SO_3^- , COO^- , or I^- as counter ion.

In the single-ion conductors described above, the ether chain dissociates counter ions from the ions attached to the

main chain and allows the ions to move much more freely. The EL intensity and the EL efficiency can be improved by employing the single-anion conductor as a hole-injecting layer or the single-cation conductor as an electron-injecting layer. However, the organic/polymer EL devices can be prepared to include either the hole-injecting layer or the electron-injecting layer to optimize the EL intensity and efficiency.

10 A preferred embodiment of the organic/polymer EL device of the present invention employing single-ion conductors is schematically depicted in FIG. 1. The organic/polymer EL device employing single-ion conductors comprises a hole-injecting layer (3) that is prepared by spin-coating of the single-anion conductor on the ITO layer prepared by depositing the semitransparent electrode material (2) on the transparent substrate (1); an emissive layer (4) prepared by spin-coating of the organic emissive material on the hole-injecting layer (3); an electron-injecting layer (5) prepared by spin-coating of the single-anion conductor on the emissive layer (4); and, a metal electrode prepared by a thermal evaporation method using the metal such as Al, Mg, Li, Ca, Au, Ag, Pt, Ni, Pb, Cu, Fe, or their alloys on the electron-injecting layer (5).

15 As described above, when single-ion conductors are used in multi-layer EL devices, the conductivity is greater than 1×10^{-8} s/cm. The EL efficiency of the device is described in quantum efficiency (% photons/electrons), which indicates the number of photons per the number of electron injected in a limit of % probability. The EL external quantum efficiency (external quantum efficiency=externally emitted photons/injected electrons $\times 100$ (%)) determined was 20 between 0.5 and 2% photons/electrons, and the turn-on voltage for the emission was as low as 1.8V.

25 The present invention is further illustrated by the following examples, which should not be taken to limit the scope 30 of the invention.

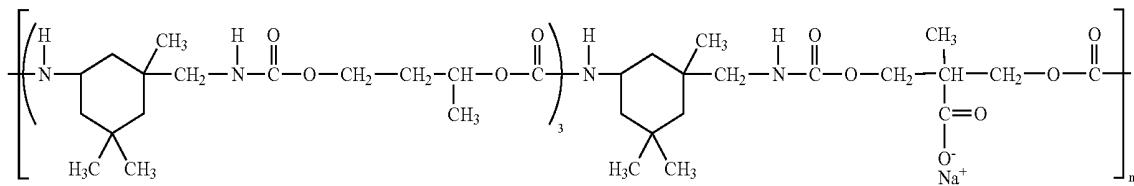
EXAMPLE 1

Preparation of an Organic/polymer EL Device Employing a Single-cation Conductor as an Electron-injecting Layer

35 Preparation of an Organic/polymer EL Device Employing a Single-cation Conductor as an Electron-injecting Layer

40 A derivative of poly(para-phenylenevinylene), MEH-PPV (poly[2-methoxy-5-(2'-ethyl-hexyl)-p-phenylenevinylene]) was spin-coated on ITO substrate in 60 nm thickness as an EL material, and then a single-cation conductor with structural formula (I) below, which has Na^+ as a counter ion by 45 ionic bond formation, was spin-coated in 15 nm thickness on the MEH-PPV layer. After that, an aluminum electrode was deposited in 100 nm thickness by a thermal evaporation method to give an organic/polymer EL device. The EL intensity was measured using a photodiode (818-UV) connected to an optical powermeter (Newport 1830-C) after applying a forward bias electric field. When EL efficiency 50 against current density of the organic/polymer EL device was calculated by measuring current while applying voltage using Keithley 236 Source measurement unit, the turn-on voltage for emission of the organic/polymer EL device was 55 60 65 70 75 80 85 90 95 100 105 110 115 120 125 130 135 140 145 150 155 160 165 170 175 180 185 190 195 200 205 210 215 220 225 230 235 240 245 250 255 260 265 270 275 280 285 290 295 300 305 310 315 320 325 330 335 340 345 350 355 360 365 370 375 380 385 390 395 400 405 410 415 420 425 430 435 440 445 450 455 460 465 470 475 480 485 490 495 500 505 510 515 520 525 530 535 540 545 550 555 560 565 570 575 580 585 590 595 600 605 610 615 620 625 630 635 640 645 650 655 660 665 670 675 680 685 690 695 700 705 710 715 720 725 730 735 740 745 750 755 760 765 770 775 780 785 790 795 800 805 810 815 820 825 830 835 840 845 850 855 860 865 870 875 880 885 890 895 900 905 910 915 920 925 930 935 940 945 950 955 960 965 970 975 980 985 990 995 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Formula I



Comparative Example 1

Preparation of an Organic/polymer EL Device
Without an Electron-injecting Layer

An organic/polymer EL device without an electron-injecting layer in Example 1, except that the spin-coating of a single-cation conductor was omitted, and EL efficiency against current was calculated.

Comparative Example 2

Preparation of an Organic/polymer EL Device
Employing an Ionomer as an Electron-injecting
Layer

An organic/polymer EL device was fabricated in a similar manner as in Example 1, except that the known electron-injecting material, a SSPS ionomer (sodium sulfonated polystyrene) was used, and then EL efficiency against current was calculated to compare with the EL efficiencies in Example 1 and Comparative Example 1 (see: FIG. 2). FIG. 2 depicts a graph comparing the EL efficiencies depending on the current densities of the organic/polymer EL devices in Example 1, Comparative Examples 1 and 2. In FIG. 2,

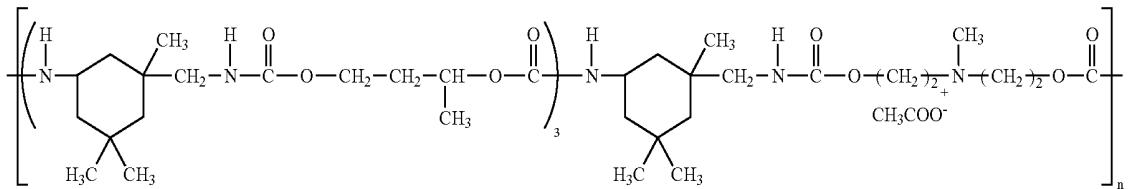
tons/electrons), and for the organic/polymer EL device employing an ionomer as an electron-injecting layer, about 0.2% (photons/electrons), and for the organic/polymer EL device without the electron-injecting layer, about 0.004% (photons/electrons), which demonstrated that the organic/polymer EL device of the present invention is highly improved in terms of the EL efficiency by employing a single-cation conductor as an electron-injecting layer.

EXAMPLE 2

Preparation of an Organic/polymer EL Device
Employing a Single-anion Conductor as a
Hole-injecting Layer (1)

A single-anion conductor with the structural formula (II) below was spin-coated in 15 nm thickness on the ITO anode substrate followed by spin-coating of the EL material, MEH-PPV in 100 nm thickness. And then, an aluminum cathode was deposited in 100 nm thickness by a thermal evaporation method to give an organic/polymer EL device. When the EL device was activated by applying a forward electric field, the turn-on voltage for emission of the organic/polymer EL device was 1.8V.

Formula II



(▲) represents the EL efficiency in case of employing a single-cation conductor as an electron-injecting layer, (●) represents the EL efficiency of the device employing an ionomer as an electron-injecting layer, and (■) represents the EL efficiency when the electron-injecting layer was not used.

As shown in FIG. 2, the EL efficiency of the invented organic/polymer EL device, employing a single-cation conductor as an electron-injecting layer, was improved by about 600 times as compared with that of not employing the electron-injecting layer, and by about 5 times compared with that of employing an ionomer as an electron-injecting layer. Further, the external quantum efficiency was calculated from the obtained results, for the invented organic/polymer EL device employing a single-cation conductor as an electron-injecting layer, which revealed that it was about 1% (photo-

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EXAMPLE 3

Preparation of an Organic/polymer EL Device
Employing a Single-anion Conductor as an
Hole-Injecting Layer (2)

An EL material, MEH-PPV was spin-coated on the ITO cathode substrate in 100 nm thickness followed by spin-coating of a single-anion conductor with the structural formula (II) above 15 nm in thickness. And then, an aluminum anode was deposited in 100 nm thickness by a thermal evaporation method to give an organic/polymer EL device. When the EL device was activated by applying reverse electric field, the turn-on voltage for emission of the organic/polymer EL device was 1.8V.

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EXAMPLE 4

Preparation of an Organic/polymer EL Device
 Employing a Single-Anion Conductor as a
 Hole-injecting Layer and a Single-cation Conductor
 as an Electron-injecting Layer

A single-anion conductor with the structural formula (II) above was spin-coated in 15 nm thickness on the ITO substrate followed by spin-coating of the EL material, MEH-PPV in 100 nm thickness. After the single-cation conductor with structural formula (I) was spin-coated in 15 nm thickness on the emissive layer, an aluminum electrode was deposited in 100 nm thickness by a thermal evaporation method to give an organic/polymer EL device. The EL intensity was measured while activating the EL device by applying forward electric fields. The turn-on voltage for emission of the organic/polymer EL device was 1.8V.

As clearly described and demonstrated as above, the present invention provides organic/polymer EL devices employing single-ion conductors as an electron- or hole-injecting layer. The organic/polymer EL device of the invention is improved in a sense that it employs an electron- or hole-injecting layer made of single-ion conductors in the EL device which comprises: a transparent substrate; a semi-transparent electrode deposited on the transparent substrate; a hole-injecting layer positioned on the semitransparent electrode; an emissive layer made of an organic emissive material, positioned on the hole-injecting layer; an electron-injecting layer positioned on the emissive layer; and, a metal electrode deposited on the electron-injecting layer. The organic/polymer EL devices of the invention have excellent EL efficiency and low turn-on voltage, which make possible their application to the development of high efficiency organic/polymer EL devices.

Although the preferred embodiments of present invention have been disclosed for illustrative purpose, those who are skilled in the art will appreciate that various modifications, additions, and substitutions are possible, without departing from the spirit and scope of the invention as disclosed in the accompanying claims.

What is claimed is:

1. An electroluminescent device comprising:
 a substrate;
 a first electrode deposited on the substrate;
 a hole-injecting layer positioned on the first electrode, the hole-injecting layer comprising a polymeric compound and a movable anion, the polymeric compound having at least one block of $[O—(CH_2)_x]_y$ units and at least one non-movable cationic moiety, wherein x is an integer equal to or greater than two, wherein y is an integer equal to or greater than one;
 an emissive layer comprising an organic electroluminescent material, positioned on the hole-injecting layer;
 an electron-injecting layer positioned on the emissive layer, the electron-injecting layer comprising a polymeric compound and a movable cation, the polymeric

compound having at least one block of $[O—(CH_2)_x]_y$ units and at least one non-movable anionic moiety, wherein x' is an integer equal to or greater than two, wherein y' is an integer equal to or greater than one; and a second electrode deposited on the electron-injecting layer.

2. An electroluminescent device comprising:
 a substrate;
 a first electrode deposited on the substrate;
 an electron-injecting layer positioned on the first electrode, the electron-injecting layer comprising a polymeric compound and a movable cation, the polymeric compound having at least one block of $[O—(CH_2)_x]_y$ units and at least one non-movable anionic moiety, wherein x' is an integer equal to or greater than two, wherein y' is an integer equal to or greater than one;
 an emissive layer comprising an organic electroluminescent material, positioned on the electron-injecting layer;
 a hole-injecting layer positioned on the emissive layer, the hole-injecting layer comprising a polymeric compound and a movable anion, the polymeric compound having at least one block of $[O—(CH_2)_x]_y$ units and at least one non-movable cationic moiety, wherein x is an integer equal to or greater than two, wherein y is an integer equal to or greater than one; and
 a second electrode deposited on the hole-injecting layer.
3. An electroluminescent device comprising:
 a substrate;
 a first electrode deposited on the substrate;
 a hole-injecting layer positioned on the first electrode, the hole-injecting layer comprising a polymeric compound and a movable anion, the polymeric compound having at least one block of $[O—(CH_2)_x]_y$ units and at least one non-movable cationic moiety, wherein x is an integer equal to or greater than two, wherein y is an integer equal to or greater than one;
 an emissive layer comprising an organic electroluminescent material, positioned on the hole-injecting layer; and,
 a second electrode deposited on the emissive layer.
4. The electroluminescent device of claim 3, wherein the substrate comprises a material selected from the group consisting of glass, quartz, and polyethylene terephthalate.
5. The electroluminescent device of claim 3, wherein the first electrode comprises a material selected from the group consisting of lead oxide, indium tin oxide, doped polyaniline, doped polypyrrole, doped polythiophene, and polyethylene dioxythiophene.
6. The electroluminescent device of claim 3, wherein the emissive layer comprises a material selected from the group consisting of emissive conjugated polymer, emissive non-conjugated polymer, emissive monomeric or oligomeric material, poly(meta-methylacrylic acid), poly(styrene), and poly(9-vinylcarbazole).

7. The electroluminescent device of claim 6, wherein the emissive conjugated polymer is selected from the group consisting of poly(p-phenylene vinylene), poly(thiophene), poly(p-phenylene), poly(fluorene), poly(arylenes), poly(arylene vinylene), polyquinoline, polypyrrole, polyaniline, polyacetylene, and derivatives thereof.

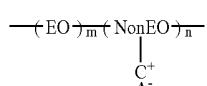
8. The electroluminescent device of claim 6, wherein the emissive non-conjugated polymer is a polymer having non-conjugated main chains and side chains substituted with emissive functional groups.

9. The electroluminescent device of claim 6, wherein the emissive monomeric or oligomeric material is selected from the group consisting of alumina quinone, rubrene, anthracene, perylene, coumarine 6, Nile red, aromatic diamine, N,N'-diphenyl-N,N'-bis-(3-methylphenyl)-1,1'-bi-phenyl-4,4'-diamine), (3-(4-biphenyl)-4-phenyl-5-(4-tert-butylphenyl)-1,2,4-triazole), (dicyanomethylene)-2-methyl-6-(p-dimethylaminostyryl)-4-H-pyran), and derivatives thereof.

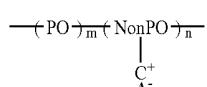
10. The electroluminescent device of claim 3, wherein the second electrode comprises a material selected from the group consisting of aluminum, magnesium, lithium, calcium, copper, silver, iron, platinum, indium, palladium, tungsten, zinc, gold, lead, and alloys thereof.

11. The electroluminescent device of claim 3, wherein the polymeric compound is represented by one or more formulas selected from the group consisting of the formula (III) and formula (IV),

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(III)

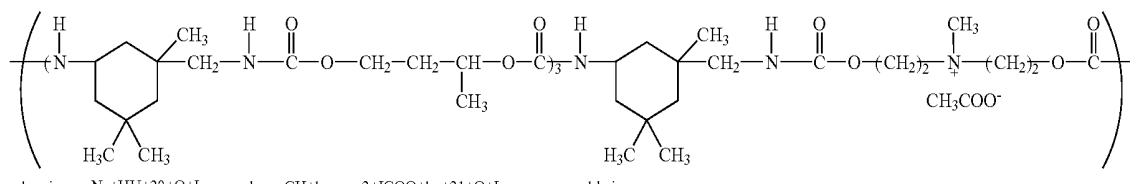
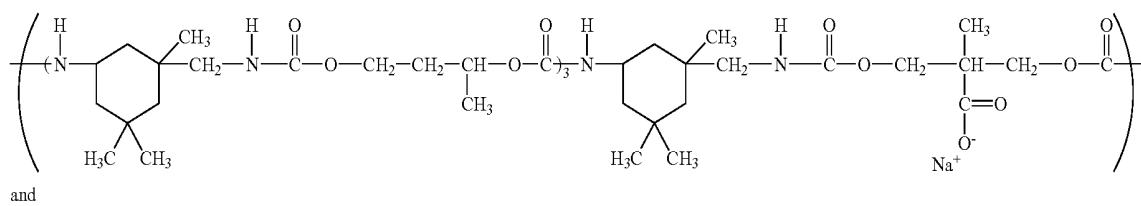


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(IV)



wherein EO represents ethylene oxide; NonEO represents non-ethylene oxide; PO represents propylene oxide; NonPO represents non-propylene oxide; A⁻ represents an anion; C⁺ represents a cation; m+n=1; and n is a number more than 0 less than 1.

45 13. The electroluminescent device of claim 3, wherein the polymeric compound comprises one or more segments selected from the group consisting of:



wherein $\text{Na}^+ \text{H}^+ + 30 + \text{O} + \text{I}^-$ and $\text{CH}_3\text{COO}^- + \text{H}^+ + 31 + \text{O} + \text{I}^-$ are movable ions.

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14. An electroluminescent device comprising:
a substrate;
a first electrode deposited on the substrate;
an emissive layer comprising an organic electrolumines-₅
cent material, positioned on the first electrode;
a hole-injecting layer positioned on the emissive layer, the
hole-injecting layer comprising a polymeric compound
and a movable anion, the polymeric compound having

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at least one block of $[O—(CH_2)_x]_y$ units and at least one
non-movable cationic moiety, wherein x is an integer
equal to or greater than two, wherein y is an integer
equal to or greater than one; and

a second electrode deposited on the hole-injecting layer.

* * * * *

专利名称(译)	采用单离子导体的有机/聚合物电致发光器件		
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申请(专利权)人(译)	PARK O型OK 李泰WOO		
当前申请(专利权)人(译)	高智控股有限责任公司81		
[标]发明人	PARK O OK LEE TAE WOO		
发明人	PARK, O-OK LEE, TAE-WOO		
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摘要(译)

本发明涉及采用单离子导体作为电子或空穴注入层材料的电致发光器件。优选的电致发光器件采用在常规电致发光器件中由单离子导体制成的电子或空穴注入层，该电致发光器件包括：透明基板；半透明电极沉积在透明基板上；位于半透明电极上的空穴注入层；由有机发光材料制成的电致发光层，位于空穴注入层上；位于电致发光层上的电子注入层；以及沉积在电子注入层上的金属电极。本发明的电致发光器件具有优异的电致发光效率和低的导通电压，这使得它们可以应用于高效电致发光器件的开发。

